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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.043$
$w R$ factor $=0.108$
Data-to-parameter ratio $=16.1$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 1-(4-Chlorobenzoyl)-2-(1H-1,2,4-triazol-1-yl)ethyl morpholine-4-carbodithioate

In the title compound, $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S}_{2}$, molecules are linked into chains along the $c$ axis by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ intermolecular interactions and further connected into a three-dimensional framework by other $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions. The packing is stabilized by $\mathrm{C}-\mathrm{H} \cdots \pi$ and $\pi-\pi$ interactions.

## Comment

Many triazole compounds have good fungicidal and plant-growth-regulating activities (Xu et al., 2002). Morpholine derivatives, an important type of fungicide, have attracted much interest because of their inward absorbent and broadspectrum activities (Badioli et al., 2001). In a search for new compounds with high bioactivity, the title compound, (I), was synthesized.

(I)

In the non-planar molecule, the orientation of the three rings is determined by the $s p^{3}$ hybridization state of atom C 8. The two aromatic rings make a dihedral angle of 81.1 (1) ${ }^{\circ}$. The morpholine ring adopts a chair conformation.

In the crystal structure, molecules are linked into chains along the $c$ axis via $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Fig. 2). The chains are further connected into a three-dimensional framework by other $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions. The packing is stabilized by $\mathrm{C}-\mathrm{H} \cdots \pi$ (Table 2) and $\pi-\pi$ interactions involving the benzene rings. The shortest distance between the centroids of neighbouring benzene rings is $3.700(2) \AA[C g \cdots C g(-x$, $1-y, 2-z$ ) where $C g$ is the centroid of the triazole ring].

## Experimental

The title compound was prepared according to a literature method (Wan et al., 2005). Single crystals suitable for an X-ray diffraction

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study were obtained by slow evaporation of an ethyl acetate-ethanol $(1: 3 \mathrm{v} / \mathrm{v})$ solution over a period of one week.

## Crystal data

$\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{ClN}_{4} \mathrm{O}_{2} \mathrm{~S}_{2}$
$M_{r}=396.91$
Monoclinic, $P 2_{1} / c$
$a=16.8253(15) \AA$
$b=7.4853(7) \AA$
$c=14.8743(13) \AA$
$\beta=97.171(1)^{\circ}$
$V=1858.7(3) \AA^{3}$
$Z=4$
$D_{x}=1.418 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 3432
reflections
$\theta=2.4-24.1^{\circ}$
$\mu=0.45 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Block, pale yellow
$0.26 \times 0.19 \times 0.05 \mathrm{~mm}$

## Data collection

Siemens SMART 1K CCD area-
detector diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.893, T_{\text {max }}=0.978$
11068 measured reflections


Figure 1
The structure of (I), showing $50 \%$ probability displacement ellipsoids and the atom-numbering scheme.


Figure 2
A view down the $b$ axis, showing the chains along the $c$ axis. Hydrogen bonds are indicated by dashed lines.
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